

***N*-[2-(6-Methyl-4-oxo-4*H*-chromen-3-yl)-4-oxothiazolidin-3-yl]furan-2-carboxamide *N,N*-dimethylformamide solvate**

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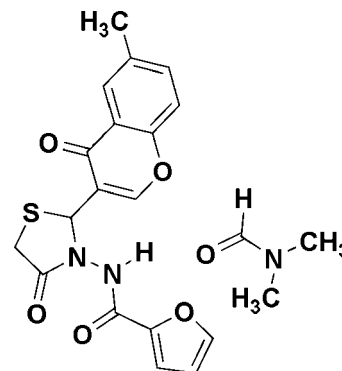
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Key indicators: single-crystal X-ray study; *T* = 292 K; mean $\sigma(\text{C}-\text{C}) = 0.003 \text{ \AA}$; *R* factor = 0.049; *wR* factor = 0.130; data-to-parameter ratio = 15.8.

The title molecule, $\text{C}_{18}\text{H}_{14}\text{N}_2\text{O}_5\text{S}\cdot\text{C}_3\text{H}_7\text{NO}$, comprises of a carboxamide group bonded to a furan ring and a distorted envelope-shaped 4-oxothiazolidin-3-yl group which is connected to a substituted 6-methyl-4-oxo-4*H*-chromen-3-yl group. Extensive strong $\text{N}-\text{H}\cdots\text{O}$ and weak $\text{C}-\text{H}\cdots\text{O}$ intermolecular hydrogen-bonding interactions occur between dimethylformamide (DMF), the crystallizing solvent, and the various heterocyclic groups within the compound, as well as additional weak $\text{C}-\text{H}\cdots\text{O}$ interactions between the heterocyclic groups themselves. The carboxyl group of the DMF solvent molecule forms a trifurcated (four-center) acceptor hydrogen-bond interaction with the carboxamide, furan and 6-methyl-4-oxo-4*H*-chromen-3-yl groups. The dihedral angles between the planar chromone group [maximum deviation = $0.0377 (18)^\circ$] and those of the furan and 4-oxothiazolidin-3-yl groups are $89.4 (6)$ and $78.5 (1)^\circ$, respectively.

Related literature

For related structures, see: Zhou *et al.* (2005). For the preparation of the title compound, see: Zhou *et al.* (2008). For general background to glycoluril and its derivatives, see: Maliar *et al.* (2004), Zhou *et al.* (2007). For puckering parameters, see: Cremer & Pople (1975).



Experimental

Crystal data

$\text{C}_{18}\text{H}_{14}\text{N}_2\text{O}_5\text{S}\cdot\text{C}_3\text{H}_7\text{NO}$
M_r = 443.48
Triclinic, *P* $\bar{1}$
a = 8.4141 (1) Å
b = 11.5676 (14) Å
c = 11.8382 (14) Å
 α = 87.138 (1) $^\circ$
 β = 70.503 (2) $^\circ$

γ = 78.419 (2) $^\circ$
V = 1063.86 (18) Å^3
Z = 2
Mo *K* α radiation
 μ = 0.20 mm^{-1}
T = 292 K
0.30 \times 0.20 \times 0.20 mm

Data collection

Bruker SMART APEX CCD area-detector diffractometer
Absorption correction: multi-scan (*SADABS*; Sheldrick, 2008)
T_{min} = 0.934, *T_{max}* = 0.962

7900 measured reflections
4568 independent reflections
3163 reflections with *I* > 2 σ (*I*)
R_{int} = 0.043

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.049$
 $wR(F^2) = 0.130$
S = 0.96
4568 reflections
289 parameters

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.25 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.17 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry (Å , $^\circ$).

<i>D</i> — <i>H</i> ··· <i>A</i>	<i>D</i> — <i>H</i>	<i>H</i> ··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> — <i>H</i> ··· <i>A</i>
N2—H2···O6	0.77 (2)	2.10 (2)	2.856 (2)	169 (2)
C4—H4···O6 ⁱ	0.93	2.50	3.421 (3)	172
C10—H10···O4 ⁱⁱ	0.93	2.49	3.332 (2)	151
C11—H11···O5 ⁱⁱ	0.98	2.50	3.267 (2)	135 (1)
C13—H13B···O3 ⁱⁱⁱ	0.97	2.55	3.441 (2)	153
C16—H16···O6	0.93	2.36	3.193 (3)	150
C18—H18···O3 ^{iv}	0.93	2.47	3.342 (3)	157
C20—H20C···O1 ^v	0.96	2.46	3.361 (3)	157

Symmetry codes: (i) $-x, -y + 1, -z + 1$; (ii) $-x, -y + 2, -z + 1$; (iii) $-x + 1, -y + 2, -z$; (iv) $-x + 1, -y + 2, -z + 1$; (v) $-x + 1, -y + 1, -z$.

Data collection: *SMART* (Bruker, 2001); cell refinement: *SAINTE* (Bruker, 2001); data reduction: *SAINTE*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *PLATON* (Spek, 2009); software used to prepare material for publication: *SHELXL97*.

The authors thank Professor Guang-Fu Yang for technical assistance and Dr Meng Xiang-Gao for the data collection. The authors acknowledge financial support from the National Natural Science Foundation of Hubei Province (grant No. 7300452) and the Medical Research Fundation of Science and Technology of Guangdong Province (grant No. B2008103)

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: JJ2002).

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Acta Cryst. (2009). E65, o2030–o2031 [doi:10.1107/S1600536809029572]

***N*-[2-(6-Methyl-4-oxo-4*H*-chromen-3-yl)-4-oxothiazolidin-3-yl]furan-2-carboxamide methylformamide solvate** *N,N*-di-

P.-L. Zhao and Z.-Z. Zhou

Comment

The tri-substituted chromone (1,4-benzopyrone) pharmacophore is an important structural element in medicinal chemistry, and shows a broad spectrum of pharmacological activities (Maliar *et al.* (2004), Zhou *et al.* (2007)). Hence, we were curious to explore the family of biheterocyclic compounds that contain both the thiazolidinone and chromone pharmacophores, with a view to discovering novel lead structures for the development of antifungal agents.

The title molecule is comprised of a carboxamide group bonded to a furan ring and a distorted envelope shaped (Cremer & Pople, 1975) 4-oxothiazolidin-3-yl group ($Q(2) = 0.1878(18) \text{ \AA}$, $\Phi(2) = 188.3(6)^\circ$; for an ideal envelope $\Phi(2) = k \times 36$) which is connected to a substituted 6-methyl-4-oxo-4*H*-chromen-3-yl group (Fig. 1). Extensive strong $N2-H2 \cdots O6$ and weak $C16-H16 \cdots O6$, $C20-H20 \cdots O6$, hydrogen bonding intermolecular interactions occur between dimethylformamide (DMF), the crystallizing solvent, and the various heterocyclic groups within the compound as well as additional weak $C-H \cdots O$ interactions between the heterocyclic groups themselves (Table 1, Fig. 2). The carboxyl group of the DMF solvent forms a trifurcated (4-center) acceptor hydrogen bond interaction with the carboxamide, furan and 6-methyl-4-oxo-4*H*-chromen-3-yl groups. The dihedral angle between the planar chromone group (max deviation = $0.0377(18)^\circ$) and that of the furan and 4-oxothiazolidin-3-yl groups is $89.4(6)^\circ$ and $78.5(1)^\circ$, respectively. Crystal packing is also stabilized by π - π stacking interactions ($Cg2-Cg2 = 3.8378(14) \text{ \AA}$; $1-x, 2-y, 1-z$; $Cg2$ is the centroid of the $O5/C15-C18$ ring).

Experimental

The title compound was synthesized according to the procedure reported (Zhou *et al.*, 2008). Crystals appropriate for X-ray data collection were obtained by slow evaporation of the DMF solution at 293 K.

Refinement

All H atoms were initially located in a difference Fourier map. The methyl H atoms were constrained to an ideal geometry, with C—H distances of 0.96 \AA , and $U_{iso}(H) = 1.49-1.50 U_{eq}C$. Each group was allowed to rotate freely about its C—C bond. All other H atoms were placed in geometrically idealized positions and constrained to ride on their parent atoms, with C—H = $0.93-0.98 \text{ \AA}$, N—H = 0.77 \AA and $U_{iso}(H) = 1.19-1.20 U_{eq}(C,N)$.

Figures



Fig. 1. The asymmetric unit of the title molecule with displacement ellipsoids for the non-hydrogen atoms drawn at the 30% probability level. Hydrogen bonds are drawn as dashed lines.

supplementary materials

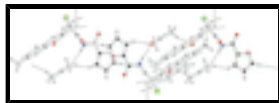


Fig. 2. The packing of the title molecule, showing one layer of molecules connected by strong N—H...O, and weak C—H...O intermolecular hydrogen bonds, and weak π - π stacking interactions.

N-[2-(6-Methyl-4-oxo-4*H*-chromen-3-yl)-4-oxothiazolidin-3-yl]furan-2-carboxamide *N,N*-dimethylformamide solvate

Crystal data

C₁₈H₁₄N₂O₅S·C₃H₇NO

M_r = 443.48

Triclinic, $P\bar{1}$

Hall symbol: -P 1

a = 8.4141 (1) Å

b = 11.5676 (14) Å

c = 11.8382 (14) Å

α = 87.138 (1)°

β = 70.503 (2)°

γ = 78.419 (2)°

V = 1063.86 (18) Å³

Z = 2

F_{000} = 464

D_x = 1.384 Mg m⁻³

Mo $K\alpha$ radiation, λ = 0.71073 Å

Cell parameters from 2421 reflections

θ = 2.5–26.7°

μ = 0.20 mm⁻¹

T = 292 K

Block, colorless

0.30 × 0.20 × 0.20 mm

Data collection

Bruker SMART APEX CCD area-detector diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

T = 292 K

φ and ω scans

Absorption correction: multi-scan (SADABS; Sheldrick, 2008)

T_{\min} = 0.934, T_{\max} = 0.962

7900 measured reflections

4568 independent reflections

3163 reflections with $I > 2\sigma(I)$

R_{int} = 0.043

θ_{max} = 27.0°

θ_{min} = 1.8°

h = -10→10

k = -13→14

l = -15→15

Refinement

Refinement on F^2

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)]$ = 0.049

$wR(F^2)$ = 0.130

S = 0.96

4568 reflections

289 parameters

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H atoms treated by a mixture of independent and constrained refinement

$$w = 1/[\sigma^2(F_o^2) + (0.0676P)^2]$$

where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} < 0.001$

$\Delta\rho_{\text{max}} = 0.25 \text{ e \AA}^{-3}$

$\Delta\rho_{\text{min}} = -0.16 \text{ e \AA}^{-3}$

Primary atom site location: structure-invariant direct methods Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.2135 (3)	0.19985 (18)	0.1714 (2)	0.0722 (7)
H1A	0.1411	0.1794	0.1309	0.108*
H1B	0.2338	0.1381	0.2251	0.108*
H1C	0.3210	0.2097	0.1135	0.108*
C2	0.1263 (3)	0.31350 (16)	0.24164 (17)	0.0519 (5)
C3	-0.0211 (3)	0.31675 (18)	0.34263 (19)	0.0580 (6)
H3	-0.0598	0.2469	0.3680	0.070*
C4	-0.1099 (3)	0.41904 (17)	0.40512 (18)	0.0542 (5)
H4	-0.2082	0.4192	0.4713	0.065*
C5	-0.0502 (2)	0.52272 (16)	0.36769 (16)	0.0445 (4)
C6	0.0963 (2)	0.52416 (15)	0.26977 (15)	0.0420 (4)
C7	0.1832 (3)	0.41704 (16)	0.20822 (16)	0.0484 (5)
H7	0.2825	0.4164	0.1427	0.058*
C8	0.1533 (2)	0.63617 (16)	0.22977 (15)	0.0427 (4)
C9	0.0433 (2)	0.73858 (15)	0.30060 (15)	0.0414 (4)
C10	-0.0936 (2)	0.72643 (16)	0.39626 (16)	0.0467 (5)
H10	-0.1573	0.7941	0.4410	0.056*
C11	0.0758 (2)	0.86128 (16)	0.26908 (16)	0.0443 (4)
H11	-0.005	0.9161	0.3323	0.053*
C12	0.3611 (3)	0.89041 (15)	0.14565 (16)	0.0469 (5)
C13	0.2766 (3)	0.90147 (18)	0.05140 (17)	0.0573 (5)
H13A	0.3236	0.8335	-0.0027	0.069*
H13B	0.2968	0.9719	0.0052	0.069*
C14	0.2690 (2)	0.93551 (16)	0.43651 (15)	0.0415 (4)
C15	0.3436 (2)	0.89915 (16)	0.53174 (15)	0.0430 (4)
C16	0.4445 (3)	0.80082 (19)	0.55371 (19)	0.0593 (5)
H16	0.4843	0.7310	0.5086	0.071*
C17	0.4781 (3)	0.8243 (2)	0.6579 (2)	0.0678 (6)
H17	0.5443	0.7728	0.6952	0.081*
C18	0.3975 (3)	0.9337 (2)	0.69290 (19)	0.0650 (6)
H18	0.3986	0.9716	0.7600	0.078*

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C19	0.7238 (4)	0.6242 (3)	0.1424 (3)	0.1048 (10)
H19A	0.6655	0.6544	0.0867	0.157*
H19B	0.8457	0.6078	0.1013	0.157*
H19C	0.6968	0.6818	0.2050	0.157*
C20	0.7468 (4)	0.4095 (3)	0.1255 (3)	0.1092 (11)
H20A	0.7150	0.3444	0.1756	0.164*
H20B	0.8696	0.4012	0.0976	0.164*
H20C	0.7074	0.4105	0.0581	0.164*
C21	0.5459 (3)	0.5218 (2)	0.2970 (2)	0.0710 (6)
H21	0.5136	0.4507	0.3256	0.085*
N1	0.2489 (2)	0.87432 (13)	0.25530 (13)	0.0442 (4)
N2	0.3049 (2)	0.85021 (14)	0.35237 (14)	0.0476 (4)
H2	0.361 (3)	0.7885 (19)	0.351 (2)	0.057*
N3	0.6696 (3)	0.51810 (17)	0.19314 (17)	0.0676 (5)
O1	0.28274 (17)	0.64100 (11)	0.14324 (11)	0.0563 (4)
O2	-0.14547 (16)	0.62410 (11)	0.43227 (11)	0.0507 (3)
O3	0.51047 (19)	0.89574 (13)	0.12709 (12)	0.0630 (4)
O4	0.18311 (18)	1.03252 (11)	0.43361 (11)	0.0534 (4)
O5	0.31260 (18)	0.98314 (12)	0.61700 (12)	0.0576 (4)
O6	0.4692 (2)	0.60997 (13)	0.35973 (14)	0.0764 (5)
S1	0.04994 (8)	0.91011 (5)	0.12526 (5)	0.0628 (2)

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.1061 (19)	0.0405 (12)	0.0708 (15)	-0.0139 (12)	-0.0301 (14)	-0.0033 (11)
C2	0.0703 (14)	0.0380 (11)	0.0530 (11)	-0.0120 (10)	-0.0274 (10)	0.0036 (9)
C3	0.0755 (15)	0.0429 (12)	0.0618 (13)	-0.0232 (11)	-0.0255 (11)	0.0115 (10)
C4	0.0597 (13)	0.0518 (12)	0.0518 (11)	-0.0213 (10)	-0.0150 (10)	0.0116 (10)
C5	0.0506 (11)	0.0418 (10)	0.0418 (10)	-0.0112 (9)	-0.0154 (8)	0.0037 (8)
C6	0.0508 (11)	0.0376 (10)	0.0395 (9)	-0.0110 (8)	-0.0163 (8)	0.0027 (8)
C7	0.0585 (12)	0.0419 (10)	0.0436 (10)	-0.0105 (9)	-0.0152 (9)	0.0012 (8)
C8	0.0508 (11)	0.0412 (10)	0.0370 (9)	-0.0129 (9)	-0.0136 (8)	0.0021 (8)
C9	0.0510 (11)	0.0369 (10)	0.0365 (9)	-0.0107 (8)	-0.0134 (8)	0.0014 (8)
C10	0.0516 (11)	0.0379 (10)	0.0460 (10)	-0.0071 (9)	-0.0109 (9)	-0.0015 (8)
C11	0.0529 (11)	0.0358 (10)	0.0403 (10)	-0.0075 (9)	-0.0110 (9)	-0.0008 (8)
C12	0.0619 (13)	0.0316 (10)	0.0436 (10)	-0.0142 (9)	-0.0098 (9)	0.0003 (8)
C13	0.0808 (15)	0.0505 (12)	0.0415 (10)	-0.0224 (11)	-0.0167 (10)	0.0060 (9)
C14	0.0460 (10)	0.0384 (10)	0.0368 (9)	-0.0152 (8)	-0.0053 (8)	-0.0005 (8)
C15	0.0456 (10)	0.0449 (11)	0.0371 (9)	-0.0163 (9)	-0.0071 (8)	-0.0004 (8)
C16	0.0659 (14)	0.0550 (13)	0.0594 (13)	-0.0103 (11)	-0.0244 (11)	-0.0015 (10)
C17	0.0738 (16)	0.0729 (16)	0.0672 (14)	-0.0154 (13)	-0.0374 (12)	0.0094 (12)
C18	0.0711 (15)	0.0889 (18)	0.0464 (12)	-0.0280 (14)	-0.0270 (11)	0.0023 (12)
C19	0.118 (3)	0.101 (2)	0.0788 (19)	-0.0156 (19)	-0.0172 (18)	0.0177 (17)
C20	0.137 (3)	0.094 (2)	0.0798 (19)	0.016 (2)	-0.0315 (19)	-0.0293 (17)
C21	0.0858 (18)	0.0557 (15)	0.0702 (16)	-0.0094 (13)	-0.0262 (14)	-0.0010 (12)
N1	0.0563 (10)	0.0408 (9)	0.0377 (8)	-0.0143 (7)	-0.0156 (7)	0.0006 (7)
N2	0.0655 (11)	0.0351 (9)	0.0426 (8)	-0.0052 (8)	-0.0206 (8)	-0.0017 (7)

N3	0.0767 (13)	0.0657 (12)	0.0566 (11)	0.0029 (10)	-0.0263 (10)	-0.0027 (10)
O1	0.0622 (9)	0.0481 (8)	0.0469 (7)	-0.0175 (7)	0.0017 (7)	-0.0035 (6)
O2	0.0532 (8)	0.0426 (8)	0.0469 (7)	-0.0117 (6)	-0.0030 (6)	0.0020 (6)
O3	0.0613 (10)	0.0649 (10)	0.0587 (9)	-0.0208 (8)	-0.0099 (7)	0.0031 (7)
O4	0.0688 (9)	0.0368 (7)	0.0531 (8)	-0.0047 (7)	-0.0206 (7)	-0.0056 (6)
O5	0.0661 (9)	0.0616 (9)	0.0465 (7)	-0.0110 (7)	-0.0197 (7)	-0.0101 (7)
O6	0.0965 (13)	0.0516 (9)	0.0663 (10)	0.0032 (9)	-0.0167 (9)	-0.0069 (8)
S1	0.0733 (4)	0.0633 (4)	0.0590 (3)	-0.0192 (3)	-0.0307 (3)	0.0198 (3)

Geometric parameters (Å, °)

C1—C2	1.508 (3)	C13—S1	1.797 (2)
C1—H1A	0.9600	C13—H13A	0.9700
C1—H1B	0.9600	C13—H13B	0.9700
C1—H1C	0.9600	C14—O4	1.212 (2)
C2—C7	1.370 (2)	C14—N2	1.354 (2)
C2—C3	1.401 (3)	C14—C15	1.470 (3)
C3—C4	1.367 (3)	C15—C16	1.347 (3)
C3—H3	0.9300	C15—O5	1.361 (2)
C4—C5	1.388 (2)	C16—C17	1.404 (3)
C4—H4	0.9300	C16—H16	0.9300
C5—O2	1.380 (2)	C17—C18	1.325 (3)
C5—C6	1.384 (2)	C17—H17	0.9300
C6—C7	1.404 (3)	C18—O5	1.363 (2)
C6—C8	1.473 (2)	C18—H18	0.9300
C7—H7	0.9300	C19—N3	1.430 (3)
C8—O1	1.230 (2)	C19—H19A	0.9600
C8—C9	1.450 (2)	C19—H19B	0.9600
C9—C10	1.344 (2)	C19—H19C	0.9600
C9—C11	1.504 (2)	C20—N3	1.438 (3)
C10—O2	1.344 (2)	C20—H20A	0.9600
C10—H10	0.9300	C20—H20B	0.9600
C11—N1	1.448 (2)	C20—H20C	0.9600
C11—S1	1.8349 (19)	C21—O6	1.224 (3)
C11—H11	0.9800	C21—N3	1.315 (3)
C12—O3	1.216 (2)	C21—H21	0.9300
C12—N1	1.356 (2)	N1—N2	1.378 (2)
C12—C13	1.498 (3)	N2—H2	0.77 (2)
C2—C1—H1A	109.5	C12—C13—H13B	110.1
C2—C1—H1B	109.5	S1—C13—H13B	110.1
H1A—C1—H1B	109.5	H13A—C13—H13B	108.4
C2—C1—H1C	109.5	O4—C14—N2	123.15 (17)
H1A—C1—H1C	109.5	O4—C14—C15	123.24 (16)
H1B—C1—H1C	109.5	N2—C14—C15	113.61 (17)
C7—C2—C3	117.65 (18)	C16—C15—O5	109.77 (17)
C7—C2—C1	122.10 (19)	C16—C15—C14	134.79 (17)
C3—C2—C1	120.22 (18)	O5—C15—C14	115.37 (16)
C4—C3—C2	122.26 (18)	C15—C16—C17	106.6 (2)
C4—C3—H3	118.9	C15—C16—H16	126.7

supplementary materials

C2—C3—H3	118.9	C17—C16—H16	126.7
C3—C4—C5	118.60 (18)	C18—C17—C16	106.9 (2)
C3—C4—H4	120.7	C18—C17—H17	126.5
C5—C4—H4	120.7	C16—C17—H17	126.5
O2—C5—C6	121.90 (15)	C17—C18—O5	110.6 (2)
O2—C5—C4	116.58 (16)	C17—C18—H18	124.7
C6—C5—C4	121.52 (18)	O5—C18—H18	124.7
C5—C6—C7	117.91 (16)	N3—C19—H19A	109.5
C5—C6—C8	120.29 (16)	N3—C19—H19B	109.5
C7—C6—C8	121.76 (16)	H19A—C19—H19B	109.5
C2—C7—C6	122.04 (18)	N3—C19—H19C	109.5
C2—C7—H7	119.0	H19A—C19—H19C	109.5
C6—C7—H7	119.0	H19B—C19—H19C	109.5
O1—C8—C9	123.67 (16)	N3—C20—H20A	109.5
O1—C8—C6	122.18 (17)	N3—C20—H20B	109.5
C9—C8—C6	114.14 (15)	H20A—C20—H20B	109.5
C10—C9—C8	120.52 (16)	N3—C20—H20C	109.5
C10—C9—C11	117.78 (16)	H20A—C20—H20C	109.5
C8—C9—C11	121.69 (15)	H20B—C20—H20C	109.5
O2—C10—C9	125.11 (17)	O6—C21—N3	126.2 (2)
O2—C10—H10	117.4	O6—C21—H21	116.9
C9—C10—H10	117.4	N3—C21—H21	116.9
N1—C11—C9	113.96 (16)	C12—N1—N2	120.45 (16)
N1—C11—S1	103.67 (11)	C12—N1—C11	120.75 (16)
C9—C11—S1	113.20 (12)	N2—N1—C11	117.91 (14)
N1—C11—H11	109.0	C14—N2—N1	119.57 (15)
C9—C11—H11	109.0	C14—N2—H2	125.4 (17)
S1—C11—H11	109.0	N1—N2—H2	114.9 (17)
O3—C12—N1	124.01 (19)	C21—N3—C19	120.1 (2)
O3—C12—C13	124.74 (17)	C21—N3—C20	121.3 (2)
N1—C12—C13	111.25 (17)	C19—N3—C20	118.6 (2)
C12—C13—S1	108.00 (13)	C10—O2—C5	117.98 (14)
C12—C13—H13A	110.1	C15—O5—C18	106.03 (16)
S1—C13—H13A	110.1	C13—S1—C11	93.51 (9)
C7—C2—C3—C4	1.8 (3)	N2—C14—C15—C16	2.3 (3)
C1—C2—C3—C4	-176.30 (19)	O4—C14—C15—O5	-1.2 (2)
C2—C3—C4—C5	-0.8 (3)	N2—C14—C15—O5	178.98 (14)
C3—C4—C5—O2	178.99 (17)	O5—C15—C16—C17	0.1 (2)
C3—C4—C5—C6	-0.3 (3)	C14—C15—C16—C17	177.0 (2)
O2—C5—C6—C7	-178.88 (16)	C15—C16—C17—C18	-0.1 (2)
C4—C5—C6—C7	0.4 (3)	C16—C17—C18—O5	0.1 (3)
O2—C5—C6—C8	-1.3 (3)	O3—C12—N1—N2	-7.0 (3)
C4—C5—C6—C8	178.01 (17)	C13—C12—N1—N2	173.40 (15)
C3—C2—C7—C6	-1.7 (3)	O3—C12—N1—C11	-175.98 (17)
C1—C2—C7—C6	176.35 (18)	C13—C12—N1—C11	4.4 (2)
C5—C6—C7—C2	0.6 (3)	C9—C11—N1—C12	109.33 (18)
C8—C6—C7—C2	-176.91 (17)	S1—C11—N1—C12	-14.16 (19)
C5—C6—C8—O1	180.00 (18)	C9—C11—N1—N2	-59.9 (2)
C7—C6—C8—O1	-2.5 (3)	S1—C11—N1—N2	176.57 (11)

C5—C6—C8—C9	-0.4 (2)	O4—C14—N2—N1	2.9 (3)
C7—C6—C8—C9	177.10 (16)	C15—C14—N2—N1	-177.28 (14)
O1—C8—C9—C10	-178.07 (19)	C12—N1—N2—C14	105.6 (2)
C6—C8—C9—C10	2.3 (3)	C11—N1—N2—C14	-85.1 (2)
O1—C8—C9—C11	3.0 (3)	O6—C21—N3—C19	-1.2 (4)
C6—C8—C9—C11	-176.62 (16)	O6—C21—N3—C20	-178.0 (2)
C8—C9—C10—O2	-2.8 (3)	C9—C10—O2—C5	1.0 (3)
C11—C9—C10—O2	176.18 (16)	C6—C5—O2—C10	1.1 (3)
C10—C9—C11—N1	127.95 (18)	C4—C5—O2—C10	-178.26 (16)
C8—C9—C11—N1	-53.1 (2)	C16—C15—O5—C18	-0.1 (2)
C10—C9—C11—S1	-113.90 (17)	C14—C15—O5—C18	-177.62 (15)
C8—C9—C11—S1	65.1 (2)	C17—C18—O5—C15	0.0 (2)
O3—C12—C13—S1	-171.45 (16)	C12—C13—S1—C11	-13.78 (14)
N1—C12—C13—S1	8.17 (19)	N1—C11—S1—C13	15.17 (13)
O4—C14—C15—C16	-177.9 (2)	C9—C11—S1—C13	-108.82 (14)

Hydrogen-bond geometry (\AA , $^\circ$)

<i>D</i> —H \cdots <i>A</i>	<i>D</i> —H	H \cdots <i>A</i>	<i>D</i> \cdots <i>A</i>	<i>D</i> —H \cdots <i>A</i>
N2—H2 \cdots O6	0.77 (2)	2.10 (2)	2.856 (2)	169 (2)
C4—H4 \cdots O6 ⁱ	0.93	2.50	3.421 (3)	172
C10—H10 \cdots O4 ⁱⁱ	0.93	2.49	3.332 (2)	151
C11—H11 \cdots O5 ⁱⁱ	0.98	2.50	3.267 (2)	135 (1)
C13—H13B \cdots O3 ⁱⁱⁱ	0.97	2.55	3.441 (2)	153
C16—H16 \cdots O6	0.93	2.36	3.193 (3)	150
C18—H18 \cdots O3 ^{iv}	0.93	2.47	3.342 (3)	157
C20—H20C \cdots O1 ^v	0.96	2.46	3.361 (3)	157

Symmetry codes: (i) $-x, -y+1, -z+1$; (ii) $-x, -y+2, -z+1$; (iii) $-x+1, -y+2, -z$; (iv) $-x+1, -y+2, -z+1$; (v) $-x+1, -y+1, -z$.

Fig. 1

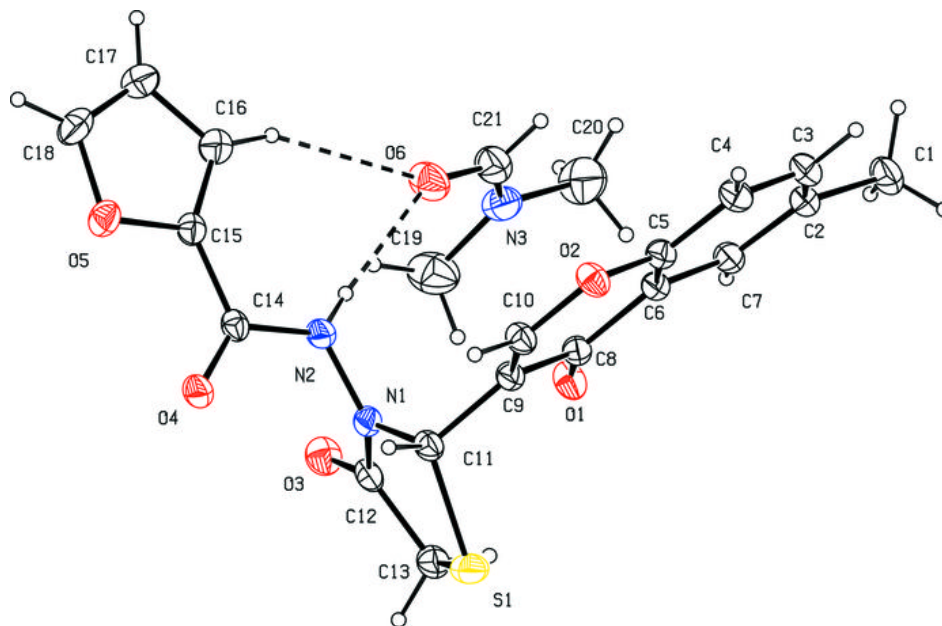


Fig. 2

